



# Experimental and Theoretical Properties of 3-Methyl/*n*-Propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-ones

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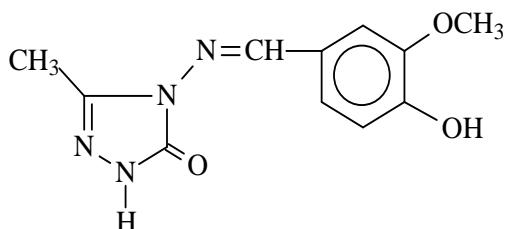
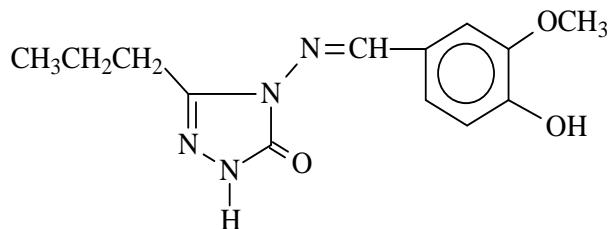
## ABSTRACT:

In this study, theoretically spectral values of 3-methyl/*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-ones were calculated and these values were compared with experimental values and obtained conclusions were evaluated. For this purpose, firstly, these compounds have been optimized using B3LYP/6-311G(d,p) and HF/6-311G(d,p) basis set. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectral values according to GIAO method was calculated using Gaussian G09W program package in gas phase and in DMSO solvent. Theoretically and experimentally values were plotted according to  $\delta_{\text{exp}} = a \cdot \delta_{\text{calc.}} + b$ , Eq. a and b constants regression coefficients with a standard error values were found using the Sigma plot program. Theoretically calculated IR values of these compounds were calculated in gas phase by using of 6-311G(d,p) basis sets of B3LYP and HF methods and are multiplied with appropriate scale factors and the values obtained according to B3LYP and HF methods are formed using theoretical infrared spectrum. The identification of calculated IR value was used veda4f program. UV-vis values in ethanol were calculated. In addition, bond angles, bond lengths, dipole moments, the highest occupied molecular orbital-lowest unoccupied molecular orbital (HOMO-LUMO) energy, mulliken charges and total energy of the molecule were calculated with both methods. The calculated and experimental results were exhibited a very good agreement.

**Key words:** 4,5-dihydro-1*H*-1,2,4-triazol-5-on, Gaussian 09W, GIAO, B3LYP, HF, 6-311G(d,p) basic set.

## 1 INTRODUCTION

The geometry of 3-methyl/*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-ones (**1** and **2**) were optimized by using the B3LYP/6-311G(d,p) and HF/6-311G(d,p) basis sets [1]. Nuclear shielding constants were also calculated by using with the same basic sets. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectrums of these compounds were studied experimentally [2] and theoretically. Afterwards, the shift values of <sup>1</sup>H NMR ve <sup>13</sup>C NMR were calculated by using the Gaussian G09W program package in gas phase and in DMSO solvent according to method of GIAO [3]. The standard error rate was calculated according to  $\delta_{\text{calc.}} = a \cdot \delta_{\text{exp.}} + b$  formula. In addition, theoretically UV-vis values in ethanol were calculated and compared with experimental values [2]. The identification of calculated IR values were used veda4f program [4]. Furthermore, the bond angles, bond lengths, dipole moments, mulliken charges, the HOMO-LUMO energy and total energy of the molecules **1** and **2** for both methods were calculated.


**1**

**2**

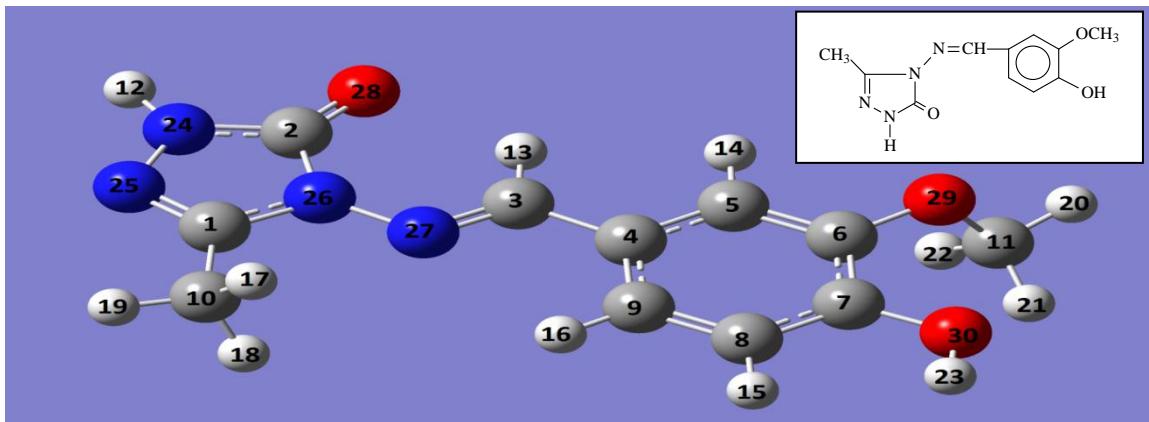
## 2 THEORETICAL

Geometry of compound **1** and **2** were optimized using B3LYP/HF methods 6-311G(d,p) basis set. The vibrational frequencies and their relative intensities were calculated by B3LYP and HF methods. The most widely used technique for calculating NMR shielding values is the GIAO (Gauge Including Atomic Orbital) method[3].This method was used for calculating <sup>1</sup>H NMR and <sup>13</sup>C NMR chemical shifts at the HF and B3LYP with the 6-311G(d,p) basis set.Time-dependent density functional theory (TD-DFT) was used to compute excitation energies and oscillator strengths for electronic transitions from ground to excited states [5-7].Some quantum chemical descriptors (bond lengths, bond angles, UV-Vis values, dipole moments, mulliken charges, HOMO-LUMO energies and total energy) are calculated at the HF/6-311G(d,p) and DFT/6-311G(d,p) level.All the calculations were carried out with the GAUSSIAN09W software [1].

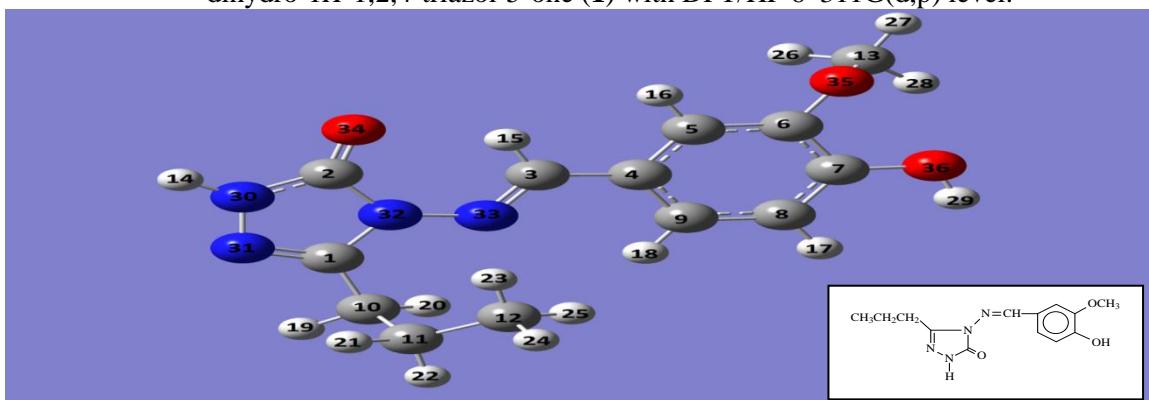
## 3 RESULTS AND DISCUSSION:

### 3.1 Molecular geometry

The optimized molecular and chemical structures of 3-methyl/n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one are shown in (Figure1 and 2) respectively. The calculated molecular geometric parameters by using the Hartree Fock (HF) and DFT (B3LYP) methods with 6-311G(d,p) basis set are given in Table 1-4. The calculated double N25=C1, N27=C3 and N31=C1, N33=C8 bond lengths according to DFT/HF methods with 6-311G(d,p) basis set of compounds **1** and **2** were found as 1.296 /1.267, 1.286 /1.258 Å<sup>0</sup> and 1.297/1.267, 1.286/1.258 Å<sup>0</sup> while the single N26-C2, N26-C1, N24-C2 and N30-C2, N32-C1, N32-C2 bond lengths in 1,2,4-triazole ring are calculated as 1.419/1.387, 1.388/1.378, 1.369/1.346 Å<sup>0</sup> and 1.368/1.346, 1.392/1.380, 1.420/1.387 Å<sup>0</sup>, respectively. The calculated double C=O bond lengths were found as 1.217/1.196 and 1.217/1.197 Å<sup>0</sup> for compounds **1** and **2** respectively. Also the single bond lengths of C6-O29, O7-O30 and C6-O35, O7-O36 were found as 1.369/1.354 1.364/1.347 and 1.369/1.354 1.364/1.347 Å<sup>0</sup> for **1** and **2** respectively. The dihedral angles of 3-methyl/n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecules is given in Table1 and3. The N27-C3-C4-C5 and N33-C3-C4-C5 dihedral angles are calculated as -179.930/-179.948 and -179.402/-178.113 Å<sup>0</sup> for compound **1** and **2** respectively. The C6-O29-C11, C7-O30-H23 and C6-O35-C11, C7-O36-H29 bond angles are calculated as 116.468/116.175, 109.109/110.634 and 116.423/116.175, 109.098/110.637 Å<sup>0</sup> for compounds **1** and **2** respectively.



**Fig.1.** The optimized molecular structure of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one (**1**) with DFT/HF 6-311G(d,p) level.



**Fig.2.** The optimized molecular structure of 3-*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one (**2**) with DFT/HF 6-311G(d,p) level.

**Table 1.** The calculated bond angles and selected dihedral angles ( $\text{A}^{\circ}$ ) of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1**)

Bond Angles ( $\text{A}^{\circ}$ )	B3LYP	HF	Selected Dihedral Angles ( $\text{A}^{\circ}$ )	B3LYP	HF
C1-N25-N24	104.680	104.976	C1-N25-N24-C2	-0.042	0.009
C1-N26-N27	121.239	121.028	N25-N24-C2-N26	0.015	-0.097
C1-N26-C2	108.254	108.060	H12-N24-C2-O28	-0.191	-0.127
C1-C10-H17	110.925	110.454	O28-C2-N26-N27	-0.165	-0.553
C1-C10-H18	110.936	110.447	N26-C1-C10-H17	59.457	59.543
C1-C10-H19	108.553	108.439	N26-C1-C10-H18	-59.797	-59.814
H17-C10-H18	107.380	107.958	N26-C1-C10-H19	179.846	179.866
H17-C10-H19	109.531	109.769	C1-N26-N27-C3	-179.408	-178.746
N25-N24-H12	120.467	121.003	N26-N27-C3-H13	0.003	-0.010
N25-N24-C2	114.413	113.703	N27-C3-C4-C5	-179.930	-179.948
N24-C2-O28	129.980	129.461	H14-C3-C4-C5	-0.078	0.122
H12-N24-C2	125.120	125.294	C3-C4-C5-C6	179.524	179.661
O28-C2-N26	128.834	128.641	C3-C4-C5-O29	-176.499	-177.690
C2-N26-N27	130.507	130.908	H14-C5-C6-O29	2.634	1.390
N26-N27-C3	118.930	119.783	C6-O29-C11-H20	176.151	178.509
N27-C3-H13	121.898	122.231	C6-O29-C11-H21	-64.697	-62.391
N27-C3-C4	120.490	120.652	C6-O29-C11-H22	57.843	59.740
H13-C3-C4	117.612	117.117	O29-C6-C7-O30	-2.266	-2.086

Bond Angles ( $\text{\AA}$ )	B3LYP	HF	Selected Dihedral Angles ( $\text{\AA}$ )	B3LYP	HF
C3-C4-C5	118.571	118.510	O29-C6-C7-C8	177.124	177.921
C3-C4-C9	122.870	122.833	C6-C7-C8-H15	179.910	-179.950
C4-C5-H14	120.665	120.649	C6-C7-C8-C9	-0.822	-0.245
C4-C5-C6	121.813	121.629	C7-C8-C9-H16	-179.529	-179.775
H14-C5-C6	117.516	117.716	C7-C8-C9-C4	0.179	0.057
C5-C6-C7	118.962	119.194	H15-C8-C9-C4	179.739	179.760
C5-C6-O29	118.836	119.767	C8-C9-C4-C5	0.559	0.198
C6-O29-C11	116.468	116.175	H16-C9-C4-C5	-179.730	-179.970
O29-C11-H20	105.847	106.557	C9-C4-C5-C6	-0.670	-0.271
O29-C11-H21	111.458	111.153			
O29-C11-H22	110.612	110.687			
H20-C11-H21	109.648	109.386			
H20-C11-H22	109.321	109.324			
H21-C11-H22	109.864	109.668			
O29-C6-C7	122.108	121.001			
C6-C7-O30	117.910	117.912			
C7-O30-H23	109.109	110.634			
O30-C7-C8	122.564	122.537			
C7-C8-H15	119.001	119.296			
C6-C7-C8	119.524	119.551			
C7-C8-C9	120.979	120.782			
H15-C8-C9	120.015	119.921			
C8-C9-H16	120.448	120.060			
C8-C9-C4	120.156	120.187			
H16-C9-C4	119.396	119.753			

**Table 2.** The calculated bond lengths ( $\text{\AA}$ ) of 3-methyl/*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1** and **2**)

Bond Lengths ( $\text{\AA}$ ) Compound 1	B3LYP	HF	Bond Lengths ( $\text{\AA}$ ) Compound 2	B3LYP	HF
C1-N26	1.388	1.378	C1-N32	1.392	1.380
C1-N25	1.296	1.267	C1-N31	1.297	1.267
C1-C10	1.485	1.487	C1-C10	1.495	1.493
C10-H17	1.093	1.084	C10-H19	1.093	1.084
C10-H18	1.093	1.084	C10-H20	1.092	1.085
C10-H19	1.089	1.081	C10-C11	1.544	1.535
N25-N24	1.380	1.369	C11-H21	1.094	1.087
N24-H12	1.006	0.990	C11-H22	1.095	1.085
N24-C2	1.369	1.346	C11-C12	1.530	1.527
C2-O28	1.217	1.196	C12-H23	1.093	1.085
C2-N26	1.419	1.387	C12-H24	1.093	1.089
N26-N27	1.372	1.365	C12-H25	1.093	1.087
N27-C3	1.286	1.258	N31-N30	1.379	1.369
C3-H13	1.087	1.075	N30-H14	1.006	0.990
C3-C4	1.462	1.472	N30-C2	1.368	1.346
C4-C5	1.402	1.390	C2-O34	1.217	1.197
C5-H14	1.084	1.075	C2-N32	1.420	1.387
C5-C6	1.389	1.379	N32-N33	1.372	1.366
C6-O29	1.369	1.354	N33-C3	1.286	1.258
O29-C11	1.433	1.409	C3-H15	1.087	1.075
C11-H20	1.089	1.083	C3-C4	1.462	1.473

Bond Lengths (Å <sup>0</sup> ) Compound 1	B3LYP	HF	Bond Lengths (Å <sup>0</sup> ) Compound 2	B3LYP	HF
C11-H21	1.092	1.087	C4-C5	1.402	1.391
C11-H22	1.096	1.083	C5-H16	1.084	1.076
C6-C7	1.407	1.391	C5-C6	1.389	1.380
C7-O30	1.364	1.347	C6-O35	1.369	1.354
C7-C8	1.398	1.387	O35-C13	1.433	1.409
O30-H23	0.963	0.941	C13-H26	1.092	1.087
C8-C9	1.386	1.380	C13-H27	1.096	1.081
C8-H15	1.087	1.077	C13-H28	1.089	1.083
C9-H16	1.082	1.073	C6-C7	1.407	1.392
			C7-O36	1.364	1.347
			C7-C8	1.398	1.387
			O36-H29	0.963	0.941
			C8-C9	1.386	1.387
			C8-H17	1.087	1.077
			C9-H18	1.082	1.073

**Table 3.** The calculated bond angles (Å<sup>0</sup>) of 3-n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**2**)

Bond Angles (Å <sup>0</sup> )	B3LYP	HF	Selected Dihedral Angles (Å <sup>0</sup> )	B3LYP	HF
C1-N30-N31	104.896	105.109	C1-N30-N31-C2	0.497	0.539
C1-N32-N33	121.801	121.273	N31-N30-C2-N32	-0.617	-0.799
C1-N32-C2	108.232	108.076	H14-N30-C2-O34	0.921	0.943
C1-C10-H19	107.111	106.475	O34-C2-N32-H14	-2.145	-2.994
C1-C10-H20	108.560	108.867	N31-C1-C10-H19	40.662	17.125
C1-C10-C11	113.803	113.130	N32-C1-C10-H19	-140.486	-164.035
H20-C10-C11	110.175	109.974	N32-C1-C10-H20	-24.093	-47.672
H19-C10-H20	108.994	108.133	C1-C10-C11-C12	-63.777	179.163
H20-C10-C11	110.175	109.974	C1-C10-C11-H21	59.276	57.712
H19-C10-C11	108.994	110.091	C1-C10-C11-H22	174.168	178.683
C10-C11-C12	114.500	111.999	H21-C11-C12-H23	-60.256	-61.144
H21-C11-C12	109.881	109.703	H22-C11-C12-H24	62.862	58.738
H22-C11-C12	109.556	109.710	H22-C11-C12-H25	-57.302	-58.585
H21-C11-H22	106.565	107.044	C1-N32-N33-C3	-178.303	-178.113
C11-C12-H23	111.616	111.172	N32-N33-C3-H15	0.169	0.008
C11-C12-H24	110.669	111.246	N33-C3-C4-C5	-179.402	-178.941
H23-C12-H24	107.521	110.879	H15-C3-C4-C5	0.727	1.195
H23-C12-H25	107.456	107.780	C3-C4-C5-C6	179.493	179.583
H24-C12-H25	107.991	107.790	C4-C5-C6-O35	-176.543	-177.664
N31-N30-H14	120.506	121.028	H16-C5-C6-O29	2.605	1.411
N31-N30-C2	114.401	113.681	C6-O35-C13-H26	58.248	59.691
N30-C2-O34	129.891	129.407	C6-O35-C13-H27	176.562	178.460
H14-N30-C2	125.073	125.265	C6-O35-C13-H28	-64.286	-62.440
O34-C2-N32	128.847	128.649	O35-C6-C7-O36	-2.243	-2.082
C2-N32-N33	129.912	130.542	O35-C6-C7-C8	177.215	177.955
N32-N33-C3	119.046	119.788	C6-C7-C8-H17	179.827	179.977
N33-C3-H15	121.945	122.243	C6-C7-C8-C9	-0.835	-0.271
N33-C3-C4	120.492	120.673	C7-C8-C9-H18	-179.720	-179.888
H15-C3-C4	117.562	117.083	C7-C8-C9-C4	0.132	0.035
C3-C4-C5	118.547	118.474	H17-C8-C9-C4	179.463	179.787
C3-C4-C9	122.922	122.888	C8-C9-C4-C5	0.646	0.277
C4-C5-H16	120.650	120.639	H18-C9-C4-C5	-179.500	-179.799

Bond Angles (A°)	B3LYP	HF	Selected Dihedral Angles (A°)	B3LYP	HF
C4-C5-C6	121.824	121.640	C9-C4-C5-C6	-0.740	-0.364
H16-C5-C6	117.521	117.715			
C5-C6-C7	118.967	119.195			
C5-C6-O35	118.867	119.769			
C6-O35-C11	116.423	116.175			
O35-C13-H26	110.610	110.686			
O35-C13-H27	105.854	106.557			
O35-C13-H28	111.447	111.150			
H26-C13-H27	109.323	109.325			
H26-C13-H28	109.649	109.387			
H27-C13-H28	109.866	109.670			
O35-C6-C7	122.074	120.998			
C6-C7-O36	117.918	117.912			
C7-O30-H29	109.098	110.637			
O36-C7-C8	122.563	122.544			
C7-C8-H17	119.000	119.297			
C6-C7-C8	119.516	120.783			
C7-C8-C9	120.975	120.783			
H17-C8-C9	120.021	119.920			
C8-C9-H18	120.386	119.920			
C8-C9-C4	120.179	120.198			

### 3.2 Vibrational frequencies

3-alkyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one (Me, *n*-Pr) molecules have 30, 36 atoms and the normal vibrational numbers are 84 and 102 respectively. The observed and calculated vibrational frequencies for compounds **1**and**2**are summarized in Table4and5. The stretching OH bands for compounds **1**and**2**were found as 3682/3763 and 3711/3763 cm<sup>-1</sup> (Fig. 3-6)at B3LYP/HF(6-31G(d,p)) level, respectively.The NH stretching bands for compounds **1**and**2**are observed at 3540/3534 and 3576/3534 cm<sup>-1</sup>, respectively. The C=O stretching vibrations of 1,2,4-triazol ring were observed at 1736/1759 and 1563/1757 cm<sup>-1</sup> for compounds **1**and **2**. The C=N stretching vibrations were found as 1588/1678 and 1505/1671 cm<sup>-1</sup>.

**Table 4.** The calculated IR frequencies of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1**).

Vibration Frequencies		Scaled DFT	Scaled HF
<b>1</b>	τ NCCC(51), τ CNNC(42),τ CCCN(18), τ NCCN(10)	<b>21</b>	<b>16</b>
<b>2</b>	δ NCNN(21), τ NNCN(16), τ COCC(13), τ CCCC(19)	<b>45</b>	<b>43</b>
<b>3</b>	τ NCC(27), δ CCC(23), τ COCC(13)	<b>59</b>	<b>58</b>
<b>4</b>	τ COCC(53), τ HCOC(15)	<b>73</b>	<b>79</b>
<b>5</b>	τ CNNC(30), τ NCNN(17), τ CNNC(12)	<b>104</b>	<b>100</b>
<b>6</b>	δ COC(13), τ HCOC(14), τ CCCC(23), τ OCCC(11)	<b>131</b>	<b>134</b>
<b>7</b>	τ HCOC(39)	<b>149</b>	<b>145</b>
<b>8</b>	τ HCCN(60)	<b>155</b>	<b>155</b>
<b>9</b>	τ CNNC(32),τ HCCN(54)	<b>163</b>	<b>170</b>
<b>10</b>	δ CCC(19), δ CNN(18)	<b>181</b>	<b>182</b>
<b>11</b>	ν CC(13), δ CCN(16)	<b>196</b>	<b>187</b>

Vibration Frequencies		Scaled DFT	Scaled HF
<b>12</b>	$\tau$ NCCC(16), $\delta$ COC(12), $\tau$ OCCC(21)	<b>214</b>	<b>198</b>
<b>13</b>	$\delta$ NCNN(19), $\tau$ CCCC(15), $\delta$ OCC(15)	<b>266</b>	<b>256</b>
<b>14</b>	$\tau$ CNNC(33), $\tau$ HNNC(20)	<b>284</b>	<b>271</b>
<b>15</b>	$\delta$ OCC(39), $\delta$ CCC(10)	<b>292</b>	<b>279</b>
<b>16</b>	$\delta$ CCN(31), $\tau$ HOCC(15)	<b>322</b>	<b>305</b>
<b>17</b>	$\tau$ HOCC(74), $\delta$ COC(22), $\tau$ CCCC(27)	<b>330</b>	<b>325</b>
<b>18</b>	$\delta$ COC(22), $\tau$ CCCC(27)	<b>349</b>	<b>353</b>
<b>19</b>	$\delta$ CNN(15), $\delta$ NNC(14), $\delta$ OCN(16), $\delta$ OCC(12)	<b>375</b>	<b>383</b>
<b>20</b>	$\tau$ CCCC(11), $\tau$ NN(17)	<b>394</b>	<b>405</b>
<b>21</b>	$\tau$ HNNC(58), $\tau$ ONNC(11)	<b>442</b>	<b>435</b>
<b>22</b>	$\delta$ CCC(39)	<b>446</b>	<b>453</b>
<b>23</b>	$\tau$ OCCC(30), $\tau$ CCCC(19), $\tau$ HCCC(19)	<b>475</b>	<b>496</b>
<b>24</b>	$\delta$ NCN(13), $\delta$ OCC(15), $\nu$ NN(12)	<b>547</b>	<b>549</b>
<b>25</b>	$\delta$ OCN(12), $\delta$ OCC(12), $\delta$ CCC(10)	<b>565</b>	<b>569</b>
<b>26</b>	$\delta$ OCN(21), $\delta$ OCC(14), $\delta$ CNN(11)	<b>586</b>	<b>591</b>
<b>27</b>	$\nu$ CC(23), $\delta$ CCC(11)	<b>615</b>	<b>616</b>
<b>28</b>	$\tau$ CCCC(28)	<b>624</b>	<b>632</b>
<b>29</b>	$\tau$ HNNC(12), $\tau$ NN(32), $\tau$ CNN(14)	<b>639</b>	<b>655</b>
<b>30</b>	$\tau$ CCCC(10), $\tau$ OCCC(25)	<b>709</b>	<b>721</b>
<b>31</b>	$\tau$ ONNC(80)	<b>716</b>	<b>748</b>
<b>32</b>	$\tau$ OCCC(14), $\nu$ CC(10), $\delta$ CCC(23)	<b>723</b>	<b>768</b>
<b>33</b>	$\delta$ CCC(13), $\nu$ OC(17)	<b>762</b>	<b>769</b>
<b>34</b>	$\delta$ CNN(25), $\nu$ NC(20)	<b>778</b>	<b>788</b>
<b>35</b>	$\tau$ HCCC(67), $\tau$ OCCC(13)	<b>783</b>	<b>823</b>
<b>36</b>	$\delta$ NCC(19), $\delta$ NNC(17), $\nu$ NN(13), $\delta$ CCC(10)	<b>820</b>	<b>834</b>
<b>37</b>	$\tau$ CCCC(13), $\tau$ HCCC(48)	<b>871</b>	<b>914</b>
<b>38</b>	$\nu$ CC(15), $\nu$ OC(16)	<b>910</b>	<b>923</b>
<b>39</b>	$\tau$ HCCC(54), $\tau$ CCCC(13)	<b>915</b>	<b>966</b>
<b>40</b>	$\delta$ NNC(13), $\tau$ HCCN(40), $\delta$ HCH(14)	<b>957</b>	<b>981</b>
<b>41</b>	$\delta$ HCNN(89)	<b>979</b>	<b>1027</b>
<b>42</b>	$\nu$ OC(59), $\delta$ CCC(17)	<b>1004</b>	<b>1036</b>
<b>43</b>	$\tau$ HCCN(13), $\delta$ HCH(20), $\tau$ HCCN(58)	<b>1028</b>	<b>1049</b>
<b>44</b>	$\delta$ NNC(26), $\nu$ NC(14), $\nu$ NN(13)	<b>1030</b>	<b>1061</b>
<b>45</b>	$\nu$ NN(26), $\delta$ HNN(11), $\tau$ HCCN(16)	<b>1062</b>	<b>1084</b>
<b>46</b>	$\delta$ CCC(10), $\nu$ OC(10), $\delta$ HOC(20), $\delta$ HCC(25)	<b>1077</b>	<b>1094</b>
<b>47</b>	$\tau$ HCOC(29), $\delta$ HCH(24)	<b>1127</b>	<b>1097</b>
<b>48</b>	$\delta$ HCC(33)	<b>1138</b>	<b>1151</b>
<b>49</b>	$\delta$ HCC(15), $\delta$ HOC(25)	<b>1144</b>	<b>1157</b>
<b>50</b>	$\tau$ HCOC(23), $\delta$ HCH(14)	<b>1162</b>	<b>1174</b>
<b>51</b>	$\nu$ NC(15), $\nu$ NN(10)	<b>1176</b>	<b>1203</b>
<b>52</b>	$\nu$ OC(32)	<b>1220</b>	<b>1208</b>

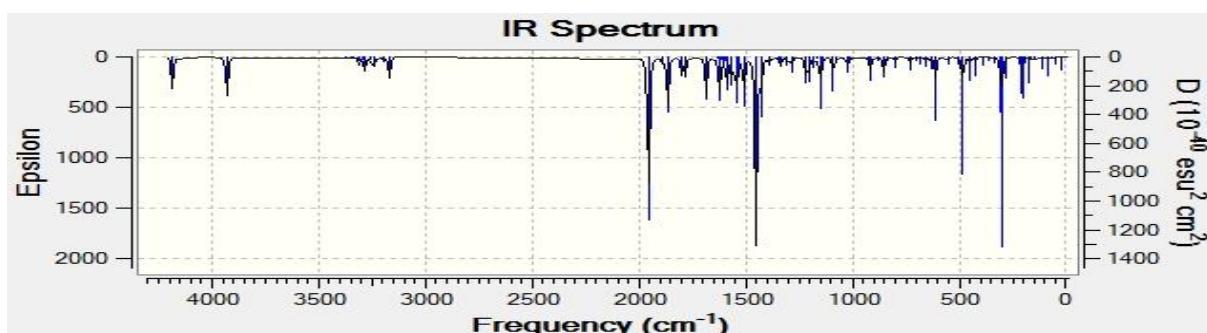
Vibration Frequencies	Scaled DFT	Scaled HF	
<b>53</b>	v NN(21), δ NCN(21), δ CNN(10)	<b>1241</b>	<b>1248</b>
<b>54</b>	δ HCC(42), v CC(16)	<b>1261</b>	<b>1285</b>
<b>55</b>	v CC(22), v OC(33)	<b>1270</b>	<b>1303</b>
<b>56</b>	δ HOC(26), δ HCC(11), v CC(38)	<b>1315</b>	<b>1306</b>
<b>57</b>	v NC(19), δ HCN(36), δ HCH(14)	<b>1328</b>	<b>1357</b>
<b>58</b>	δ HNN(61), δ HCH(47)	<b>1343</b>	<b>1380</b>
<b>59</b>	v CC(10), δ HCN(13), δ HCH(52)	<b>1364</b>	<b>1390</b>
<b>60</b>	v CC(24), δ HCH(18)	<b>1384</b>	<b>1407</b>
<b>61</b>	δ HCN(16), δ HCH(40)	<b>1404</b>	<b>1428</b>
<b>62</b>	δ HCH(78), τ HCCN(21)	<b>1416</b>	<b>1431</b>
<b>63</b>	δ HCH(42)	<b>1426</b>	<b>1446</b>
<b>64</b>	δ HCH(45), τ HCOC(17)	<b>1430</b>	<b>1451</b>
<b>65</b>	τ HCCN(13), δ HCH(34)	<b>1436</b>	<b>1461</b>
<b>66</b>	δ HCH(51), τ HCOC(21)	<b>1454</b>	<b>1465</b>
<b>67</b>	v OC(15), δ HCC(39), δ CCC(12), v CC(11)	<b>1488</b>	<b>1516</b>
<b>68</b>	v CC(13), δ CCC(19)	<b>1559</b>	<b>1604</b>
<b>69</b>	v CC(42)	<b>1579</b>	<b>1617</b>
<b>70</b>	v NC(53)	<b>1588</b>	<b>1678</b>
<b>71</b>	v NC(58), v CC(11)	<b>1602</b>	<b>1697</b>
<b>72</b>	v OC(73), v NC(13)	<b>1736</b>	<b>1759</b>
<b>73</b>	v CH(91)	<b>2899</b>	<b>2849</b>
<b>74</b>	v CH(93)	<b>2929</b>	<b>2872</b>
<b>75</b>	v CH(70)	<b>2974</b>	<b>2916</b>
<b>76</b>	v CH(100)	<b>2980</b>	<b>2928</b>
<b>77</b>	v CH(71)	<b>3013</b>	<b>2951</b>
<b>78</b>	v CH(93)	<b>3021</b>	<b>2959</b>
<b>79</b>	v CH(34)	<b>3024</b>	<b>2977</b>
<b>80</b>	v CH(42)	<b>3031</b>	<b>3003</b>
<b>81</b>	v CH(40)	<b>3062</b>	<b>3014</b>
<b>82</b>	v CH(33)	<b>3083</b>	<b>3031</b>
<b>83</b>	v NH(100)	<b>3540</b>	<b>3534</b>
<b>84</b>	v OH(100)	<b>3682</b>	<b>3763</b>

**Table 5.** The calculated IR frequencies of 3-*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**2**)

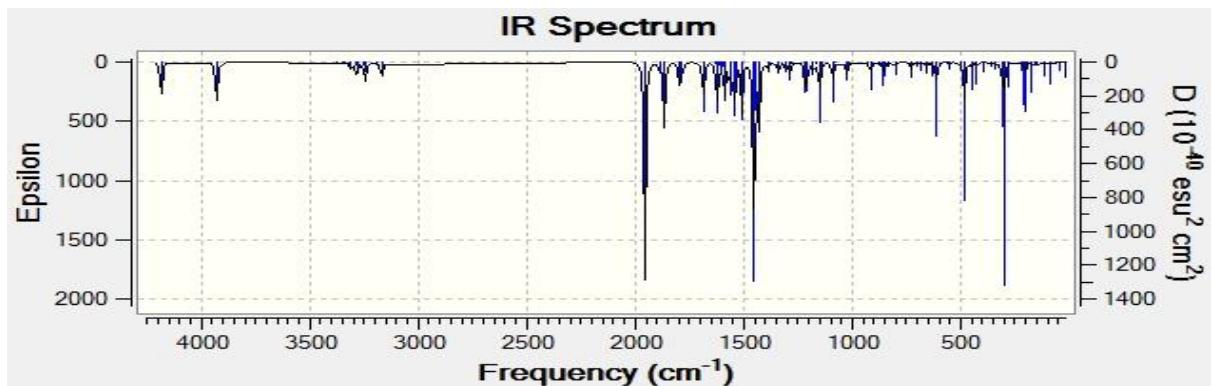
Vibration Frequencies		Scaled DFT	Scaled HF
<b>1</b>	$\tau$ CNNC (39), $\tau$ CCCN(17), $\tau$ NCCN(11)	<b>13</b>	<b>13</b>
<b>2</b>	$\tau$ CCCN(14), $\tau$ CCNN(13), $\tau$ NCCN(13)	<b>33</b>	<b>34</b>
<b>3</b>	$\delta$ NCC(18), $\tau$ NNCN(11), $\delta$ CCC(19)	<b>50</b>	<b>48</b>
<b>4</b>	$\tau$ COCC(27)	<b>64</b>	<b>61</b>
<b>5</b>	$\delta$ CCC(11), $\tau$ CNNC(15), $\tau$ CCCN(20)	<b>77</b>	<b>76</b>
<b>6</b>	$\tau$ COCC(34), $\tau$ CCCN(12)	<b>85</b>	<b>83</b>
<b>7</b>	$\tau$ COCC(12), $\tau$ CCCN(27), $\tau$ HCCN(10)	<b>91</b>	<b>89</b>
<b>8</b>	$\tau$ CNNC(29), $\tau$ CCCN(11), $\tau$ NCCN(10)	<b>115</b>	<b>113</b>
<b>9</b>	$\tau$ CCCN(24), $\tau$ OCCC(10), $\delta$ COC(14)	<b>116</b>	<b>135</b>
<b>10</b>	$\tau$ HCOC(39)	<b>117</b>	<b>146</b>
<b>11</b>	$\tau$ HOCC(43), $\tau$ CNNC(17), $\tau$ CCCN(21), $\delta$ CCC(11)	<b>118</b>	<b>182</b>
<b>12</b>	$\tau$ CCCN(19), $\tau$ OCCC(18), $\delta$ COC(10)	<b>119</b>	<b>185</b>
<b>13</b>	$\tau$ HOCC(49), $\tau$ CCCN(17), $\tau$ NCNN(18), $\delta$ CCN(11)	<b>143</b>	<b>191</b>
<b>14</b>	$\tau$ NNCN(15), $\tau$ COCC(30), $\delta$ CCN(24), $\tau$ CCCC(13)	<b>178</b>	<b>202</b>
<b>15</b>	$\nu$ CC(21), $\tau$ HCCC(71)	<b>213</b>	<b>238</b>
<b>16</b>	$\tau$ CCCC(14), $\delta$ CNN(18), $\delta$ CCC(11)	<b>231</b>	<b>256</b>
<b>17</b>	$\delta$ CCC(10), $\tau$ HOCC(88)	<b>284</b>	<b>271</b>
<b>18</b>	$\delta$ CCN(11), $\delta$ OCC(11)	<b>294</b>	<b>292</b>
<b>19</b>	$\delta$ OCC(12), $\delta$ COC(16)	<b>308</b>	<b>302</b>
<b>20</b>	$\tau$ CCCC(22), $\tau$ NNCN(10)	<b>346</b>	<b>322</b>
<b>21</b>	$\tau$ HCOC(57), $\tau$ COCC(20), $\delta$ OCC(12)	<b>380</b>	<b>339</b>
<b>22</b>	$\delta$ CCC(23)	<b>395</b>	<b>361</b>
<b>23</b>	$\tau$ CCCC(16), $\tau$ NNCN(11)	<b>417</b>	<b>384</b>
<b>24</b>	$\delta$ CNN(11), $\delta$ CCC(17), $\nu$ NC(11)	<b>432</b>	<b>410</b>
<b>25</b>	$\tau$ CCCC(13), $\tau$ OCCC(31), $\tau$ HCCC(18)	<b>454</b>	<b>451</b>
<b>26</b>	$\delta$ CCC(34)	<b>472</b>	<b>454</b>
<b>27</b>	$\nu$ CC(25), $\delta$ NNC(10), $\tau$ HCCN(11)	<b>509</b>	<b>497</b>
<b>28</b>	$\delta$ OCC(24), $\delta$ CCC(11)	<b>532</b>	<b>554</b>
<b>29</b>	$\delta$ CCC(13), $\nu$ OC(11), $\delta$ OCN(10)	<b>547</b>	<b>571</b>
<b>30</b>	$\delta$ NCC(15), $\delta$ OCN(25)	<b>595</b>	<b>593</b>
<b>31</b>	$\tau$ CCCC(22), $\tau$ HCCC(19), $\tau$ OCCC(20)	<b>611</b>	<b>630</b>
<b>32</b>	$\nu$ NC(16)	<b>660</b>	<b>638</b>
<b>33</b>	$\delta$ OCC(10), $\delta$ COC(15), $\tau$ CCCC(20), $\tau$ HCCC(11)	<b>678</b>	<b>706</b>
<b>34</b>	$\delta$ CCC(10)	<b>690</b>	<b>718</b>
<b>35</b>	$\tau$ OCCC(14), $\tau$ HCCC(14)	<b>692</b>	<b>734</b>
<b>36</b>	$\tau$ HCCN(11)	<b>702</b>	<b>749</b>
<b>37</b>	$\delta$ COC(14), $\nu$ OC(18), $\delta$ CCC(18), $\nu$ CC(12)	<b>711</b>	<b>768</b>
<b>38</b>	$\tau$ OCCC(20), $\tau$ HCCC(31)	<b>736</b>	<b>773</b>
<b>39</b>	$\nu$ CC(10), $\delta$ NNC(12), $\delta$ CNN(16), $\nu$ NC(13)	<b>749</b>	<b>803</b>
<b>40</b>	$\nu$ OC(15)	<b>771</b>	<b>823</b>

Vibration Frequencies	Scaled DFT	Scaled HF	
<b>41</b>	$\tau$ HCCC(32), $\tau$ CCCC(11)	<b>804</b>	<b>834</b>
<b>42</b>	$\delta$ OCN(14), $\tau$ ONNC(23)	<b>822</b>	<b>855</b>
<b>43</b>	$\tau$ HCCN(18), $\nu$ CC(49)	<b>836</b>	<b>868</b>
<b>44</b>	$\nu$ CC(12), $\nu$ NC(10)	<b>849</b>	<b>914</b>
<b>45</b>	$\tau$ HCCC(68)	<b>871</b>	<b>923</b>
<b>46</b>	$\nu$ CC(35), $\tau$ HCCC(81)	<b>887</b>	<b>965</b>
<b>47</b>	$\nu$ NC(14)	<b>904</b>	<b>993</b>
<b>48</b>	$\tau$ HCNN(79)	<b>928</b>	<b>1025</b>
<b>49</b>	$\nu$ NN(37)	<b>963</b>	<b>1029</b>
<b>50</b>	$\nu$ OC(57), $\delta$ HCC(10)	<b>1008</b>	<b>1036</b>
<b>51</b>	$\nu$ OC(23), $\nu$ CC(13)	<b>1012</b>	<b>1068</b>
<b>52</b>	$\nu$ OC(37)	<b>1016</b>	<b>1083</b>
<b>53</b>	$\nu$ CC(32), $\tau$ HCCC(15)	<b>1019</b>	<b>1085</b>
<b>54</b>	$\nu$ NN(45)	<b>1043</b>	<b>1094</b>
<b>55</b>	$\nu$ CC(28), $\tau$ HCCC(22), $\tau$ HCCN(23)	<b>1059</b>	<b>1117</b>
<b>56</b>	$\delta$ HOC(21), $\delta$ HCC(16)	<b>1068</b>	<b>1151</b>
<b>57</b>	$\delta$ HCC(21), $\tau$ HCOC(15), $\delta$ HCH(11)	<b>1091</b>	<b>1157</b>
<b>58</b>	$\delta$ HCC(30)	<b>1093</b>	<b>1174</b>
<b>59</b>	$\nu$ NN(16), $\delta$ CNN(12)	<b>1108</b>	<b>1203</b>
<b>60</b>	$\delta$ CCC(11), $\delta$ HOC(21)	<b>1115</b>	<b>1209</b>
<b>61</b>	$\delta$ HCC(42)	<b>1126</b>	<b>1227</b>
<b>62</b>	$\tau$ HCOC(19)	<b>1130</b>	<b>1248</b>
<b>63</b>	$\nu$ CC(28), $\nu$ OC(27)	<b>1145</b>	<b>1272</b>
<b>64</b>	$\delta$ HNN(19), $\delta$ HCC(34), $\tau$ HCCN(11)	<b>1174</b>	<b>1285</b>
<b>65</b>	$\delta$ HNN(33), $\delta$ HCC(51), $\tau$ HCCN(15)	<b>1182</b>	<b>1289</b>
<b>66</b>	$\delta$ CNN(13), $\tau$ HCCN(17)	<b>1209</b>	<b>1305</b>
<b>67</b>	$\delta$ HCC(45)	<b>1221</b>	<b>1314</b>
<b>68</b>	$\delta$ HCC(33), $\tau$ HCCN(22)	<b>1241</b>	<b>1359</b>
<b>69</b>	$\tau$ HCCN(41)	<b>1267</b>	<b>1370</b>
<b>70</b>	$\nu$ CC(24)	<b>1269</b>	<b>1384</b>
<b>71</b>	$\delta$ HCH(38)	<b>1314</b>	<b>1391</b>
<b>72</b>	$\delta$ HCN(62)	<b>1323</b>	<b>1404</b>
<b>73</b>	$\delta$ CCC(10), $\delta$ OCC(10), $\nu$ CC(16)	<b>1361</b>	<b>1435</b>
<b>74</b>	$\delta$ HCH(28)	<b>1388</b>	<b>1441</b>
<b>75</b>	$\delta$ HCH(43), $\tau$ HCOC(10)	<b>1402</b>	<b>1450</b>
<b>76</b>	$\delta$ HCH(37), $\tau$ HCCC(12)	<b>1408</b>	<b>1451</b>
<b>77</b>	$\delta$ HCH(43), $\tau$ HCOC(11)	<b>1418</b>	<b>1452</b>
<b>78</b>	$\delta$ HCH(52)	<b>1422</b>	<b>1456</b>
<b>79</b>	$\delta$ HCC(14), $\nu$ CC(16)	<b>1428</b>	<b>1463</b>
<b>80</b>	$\delta$ HCH(43), $\tau$ HCCC(18)	<b>1442</b>	<b>1465</b>

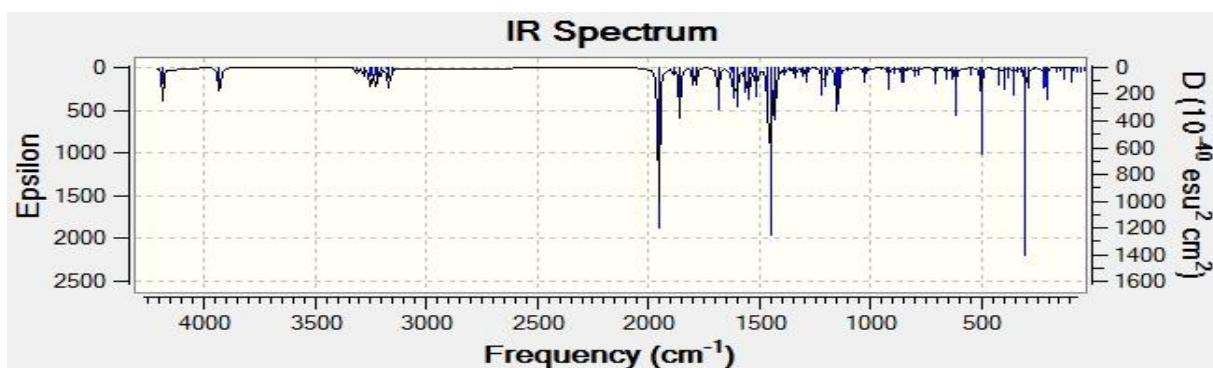
Vibration Frequencies	Scaled DFT	Scaled HF	
<b>81</b>	v OC(44), v NC(19)	<b>1458</b>	<b>1516</b>
<b>82</b>	v OC(23), v NC(45)	<b>1467</b>	<b>1604</b>
<b>83</b>	v NC(32)	<b>1505</b>	<b>1617</b>
<b>84</b>	v NC(38), v CC(23), δ HCH(14)	<b>1519</b>	<b>1671</b>
<b>85</b>	v CC(26), δ HCH(16)	<b>1544</b>	<b>1696</b>
<b>86</b>	v OC(17)	<b>1563</b>	<b>1757</b>
<b>87</b>	v CH(94)	<b>3091</b>	<b>2838</b>
<b>88</b>	v CH(98)	<b>3093</b>	<b>2849</b>
<b>89</b>	v CH(56)	<b>3116</b>	<b>2855</b>
<b>90</b>	v CH(56)	<b>3128</b>	<b>2879</b>
<b>91</b>	v CH(58)	<b>3168</b>	<b>2880</b>
<b>92</b>	v CH(53)	<b>3178</b>	<b>2898</b>
<b>93</b>	v CH(88)	<b>3179</b>	<b>2901</b>
<b>94</b>	v CH(88)	<b>3179</b>	<b>2917</b>
<b>95</b>	v CH(81)	<b>3182</b>	<b>2927</b>
<b>96</b>	v CH(89)	<b>3186</b>	<b>2951</b>
<b>97</b>	v CH(66)	<b>3190</b>	<b>2977</b>
<b>98</b>	v CH(41)	<b>3190</b>	<b>3004</b>
<b>99</b>	v CH(67)	<b>3194</b>	<b>3014</b>
<b>100</b>	v CH(35)	<b>3207</b>	<b>3031</b>
<b>101</b>	v NH(100)	<b>3576</b>	<b>3534</b>
<b>102</b>	v OH(100)	<b>3711</b>	<b>3763</b>



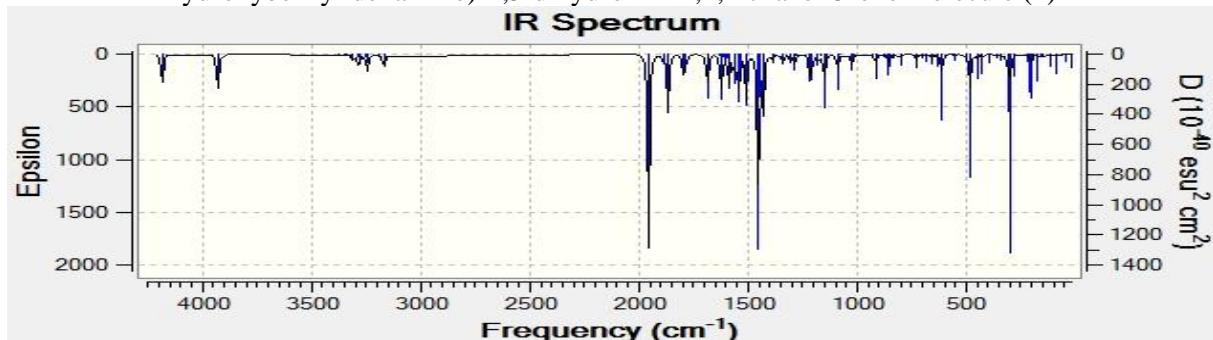
**Fig.3.** IR spectrum simulated DFT/6-311G(d) level of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1**)



**Fig.4.** IR spectrum simulated HF/6-311G(d) level of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**1**)



**Fig.5.** IR spectrum simulated HF/6-311G(d) level of 3-n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**2**)



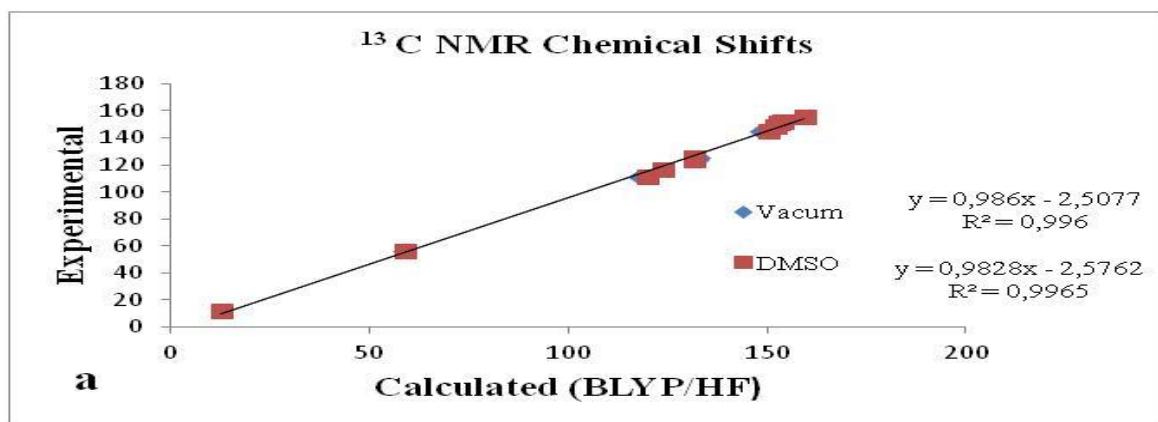
**Fig.6.** IR spectrum simulated HF/6-311G(d) level of 3-n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**2**)

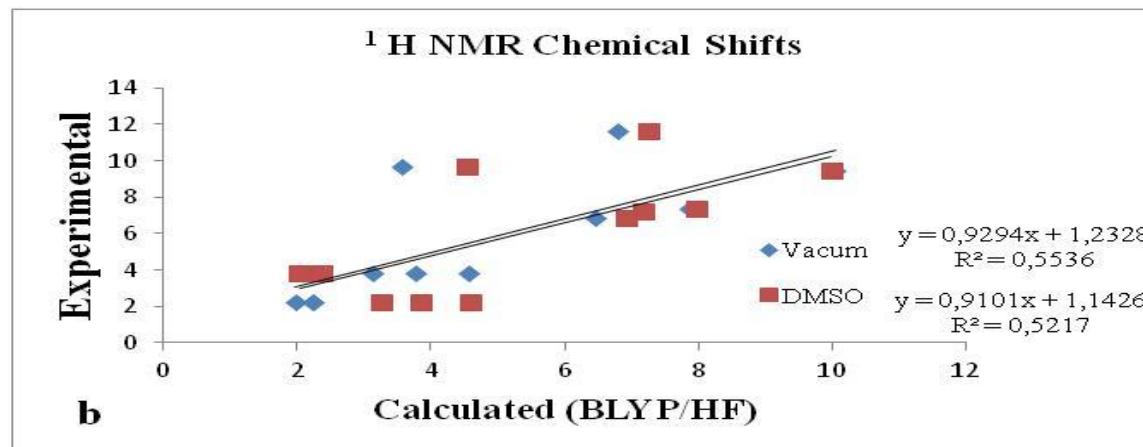
### 3.3 $^{13}\text{C}$ and $^1\text{H}$ NMR Chemical Shifts and Regression Analyses

The  $^1\text{H}$  and  $^{13}\text{C}$  NMR chemical shifts of the title compounds in gase phase and in DMSO solvent have been calculated by using the DFT (B3LYP) and Hartree Fock (HF) methods with 6-311G(d,p) basis set. The  $R^2$  values of compounds **1** and **2** were evaluated and  $^{13}\text{C}$  and  $^1\text{H}$  NMR chemical shift values of the compounds **1** and **2** were plotted graphics (Fig. 7 and 8).  $^{13}\text{C}$  and  $^1\text{H}$  NMR chemical shift values of the compounds **1** and **2** given in Table 5 and 6. Theoretical and experimental between  $^{13}\text{C}$  and  $^1\text{H}$  NMR chemical shifts ratios of compounds **1** and **2** were observed a linear correlation except for NH proton by  $R^2$  because N-H proton of 4,5-dihydro-1H-1,2,4-triazol-5-one ring was displayed the acidic character (Fig. 7 and 8). The calculated  $^{13}\text{C}$  and  $^1\text{H}$  NMR chemical shifts values for the mentioned compounds are in a very good agreement with the experimental data [1].

**Table 5.** The calculated  $^{13}\text{C}$  and  $^1\text{H}$  NMR isotropic chemical shifts of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (1, 2) (with respect to TMS, all values in ppm).

Nucleus	$\delta_{\text{Exp.}}$	$\delta_{\text{cal.DFT}}$	Different	$\delta_{\text{cal.DFT}}(\text{DMSO})$	Different	$\delta_{\text{cal.HF(Vacuum)}}$	Different	$\delta_{\text{cal.HF(DMSO)}}$	Different
<b>C1</b>	144,86	148,30	-3,44	150,26	-5,40	139,02	5,84	141,47	3,39
<b>C2</b>	152,05	152,72	-0,67	153,76	-1,71	141,92	10,13	142,94	9,11
<b>C3</b>	150,85	152,53	-1,68	153,13	-2,28	143,76	7,09	144,56	6,29
<b>C4</b>	125,47	132,95	-7,48	131,59	-6,12	116,82	8,65	115,71	9,76
<b>C5</b>	123,23	132,38	-9,15	131,70	-8,47	123,69	-0,46	123,35	-0,12
<b>C6</b>	148,71	152,72	-4,01	152,09	-3,38	135,38	13,33	134,35	14,36
<b>C7</b>	155,41	158,23	-2,82	159,32	-3,91	143,66	11,75	144,21	11,20
<b>C8</b>	110,91	117,57	-6,66	119,81	-8,90	104,14	6,77	106,03	4,88
<b>C9</b>	116,31	123,49	-7,18	123,93	-7,62	116,57	-0,26	117,25	-0,94
<b>C10</b>	11,81	12,72	-0,91	12,72	-0,91	0,82	10,99	0,90	10,91
<b>C11</b>	56,28	58,66	-2,38	59,04	-2,76	41,68	14,60	42,04	14,24
<b>H12</b>	11,64	6,80	4,84	7,26	4,38	6,04	5,60	6,45	5,19
<b>H13</b>	9,49	10,06	-0,57	9,99	-0,50	9,23	0,26	9,19	0,30
<b>H14</b>	7,23	7,15	0,08	7,17	0,06	7,10	0,13	7,19	0,04
<b>H15</b>	6,86	6,47	0,39	6,92	-0,06	6,17	0,69	6,63	0,23
<b>H16</b>	7,37	7,88	-0,51	7,97	-0,60	7,75	-0,38	7,88	-0,51
<b>H17</b>	2,24	2,24	0,00	3,84	-1,60	1,79	0,45	1,94	0,30
<b>H18</b>	2,24	1,97	0,27	4,60	-2,36	1,64	0,60	1,95	0,29
<b>H19</b>	2,24	2,24	0,00	3,25	-1,01	1,80	0,44	1,69	0,55
<b>H20</b>	3,81	3,77	0,04	2,36	1,45	3,28	0,53	3,27	0,54
<b>H21</b>	3,81	3,12	0,69	2,36	1,45	3,25	0,56	3,30	0,51
<b>H22</b>	3,81	4,56	-0,75	2,03	1,78	2,43	1,38	2,62	1,19
<b>H23</b>	9,65	3,56	6,09	4,53	5,12	5,65	4,00	3,54	6,11

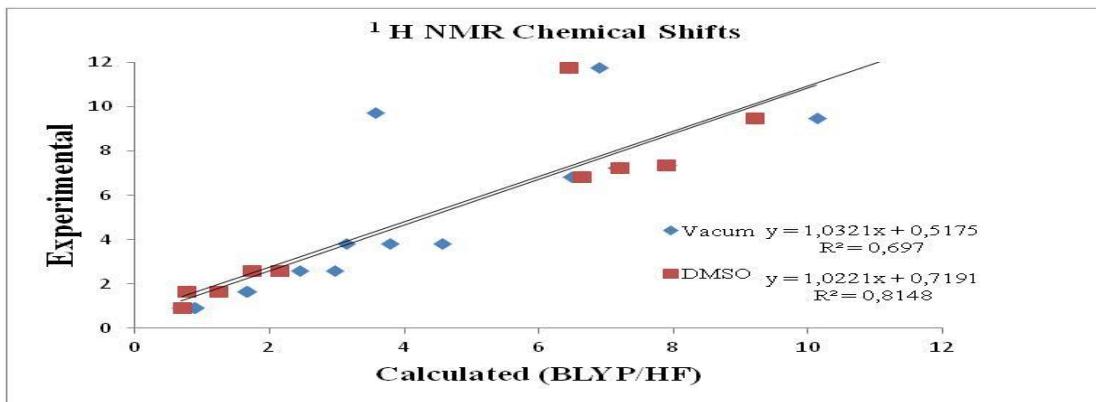




**Fig.7.** The correlation <sup>13</sup>C (a) and <sup>1</sup>H (b) NMR isotropic chemical shifts of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**1**)

**Table 6.** The calculated and experimental <sup>1</sup>H NMR isotropic chemical shifts of 3-n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**2**) (with respect to TMS, allvalues in ppm).

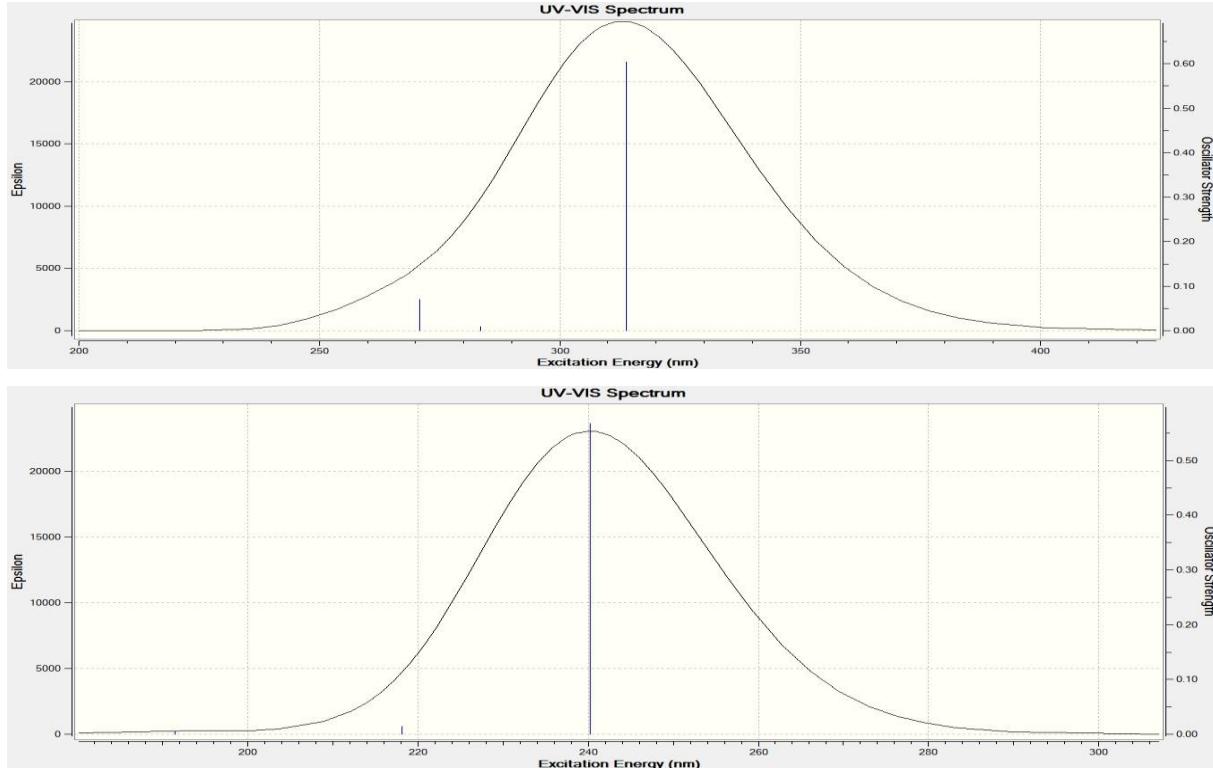
Nucleus	$\delta_{\text{Exp.}}$	$\delta_{\text{cal.DFT}}(\text{Vacum})$	Different	$\delta_{\text{cal.DFT}}(\text{DMSO})$	Different	$\delta_{\text{cal.HF}}(\text{Vacum})$	Different	$\delta_{\text{cal.HF}}(\text{DMSO})$	Different
<b>H14</b>	11,78	6,90	4,88	6,45	5,33	6,73	5,05	7,38	4,40
<b>H15</b>	9,47	10,13	-0,66	9,2	0,27	9,96	-0,49	10,06	-0,59
<b>H16</b>	7,23	7,16	0,07	7,2	0,03	7,82	-0,59	7,18	0,05
<b>H17</b>	6,85	6,48	0,37	6,64	0,21	6,90	-0,05	6,93	-0,08
<b>H18</b>	7,36	7,91	-0,55	7,89	-0,53	8,46	-1,10	8,02	-0,66
<b>H19</b>	2,60	2,97	-0,37	2,15	0,45	2,73	-0,13	3,04	-0,44
<b>H20</b>	2,60	2,45	0,15	1,73	0,87	2,40	0,20	2,53	0,07
<b>H21</b>	1,65	1,66	-0,01	1,24	0,41	1,99	-0,34	1,81	-0,16
<b>H22</b>	1,65	1,65	0,00	0,75	0,90	1,46	0,19	1,59	0,06
<b>H23</b>	0,92	0,68	0,24	0,7	0,22	1,41	-0,49	0,91	0,01
<b>H24</b>	0,92	0,86	0,06	0,59	0,33	1,21	-0,29	0,98	-0,06
<b>H25</b>	0,92	0,89	0,03	0,51	0,41	1,19	-0,27	0,62	0,30
<b>H26</b>	3,81	3,13	0,68	3,3	0,51	3,92	-0,11	4,59	-0,78
<b>H27</b>	3,81	3,78	0,03	3,27	0,54	3,97	-0,16	3,85	-0,04
<b>H28</b>	3,81	4,55	-0,74	2,62	1,19	3,15	0,66	3,27	0,54
<b>H29</b>	9,74	3,56	6,18	3,54	6,20	3,37	6,37	4,53	5,21



**Fig.8.** The correlation  $^1\text{H}$  NMR isotropic chemical shifts of 3-*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**2**)

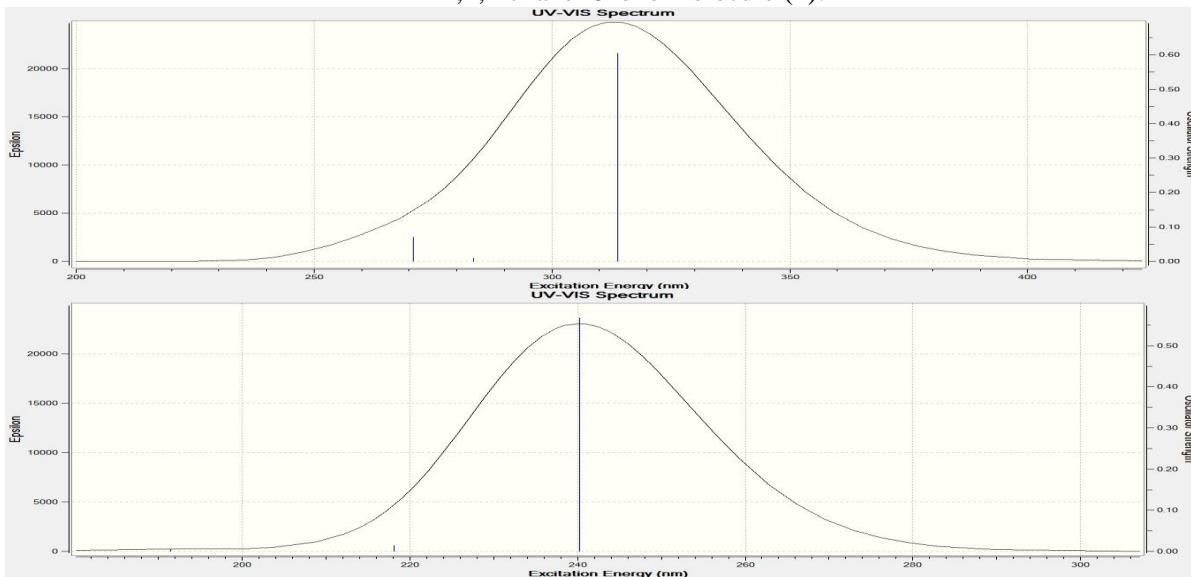
### 3.4 UV-vis. Spectroscopy and HOMO-LUMO Analyses

The experimental absorption wavelenghts [1] of the compounds **1** and **2** in ethanol solvent have been observed 332,308, 234,220 nm and 312, 286, 234, 216 nm respectively. The excitation energies, oscillator strengths (*f*) and absorption wavelengths ( $\lambda$ ) of UV–Vis electron absorption spectroscopy [8] of the title molecules have been calculated in ethanol solvent by using B3LYP/HF methods with 6-311G(d,p) basis set and are presented in Figure 9 and 10. The calculated electron absorption wavelengths for the compounds **1** and **2** were found as 313,74/240,30283,46/218,12, 270,86/191,49 and 314,16/223,27, 284,08/242,45, 271,44/194,15 nm in ethanol solvent, respectively. Furthermore, the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO)have been simulated for compounds **1** and **2**have been determined. In our study, HOMO and LUMO energies and their 3D plots of the compounds **1** and **2**are shown in Fig 11 and 12.



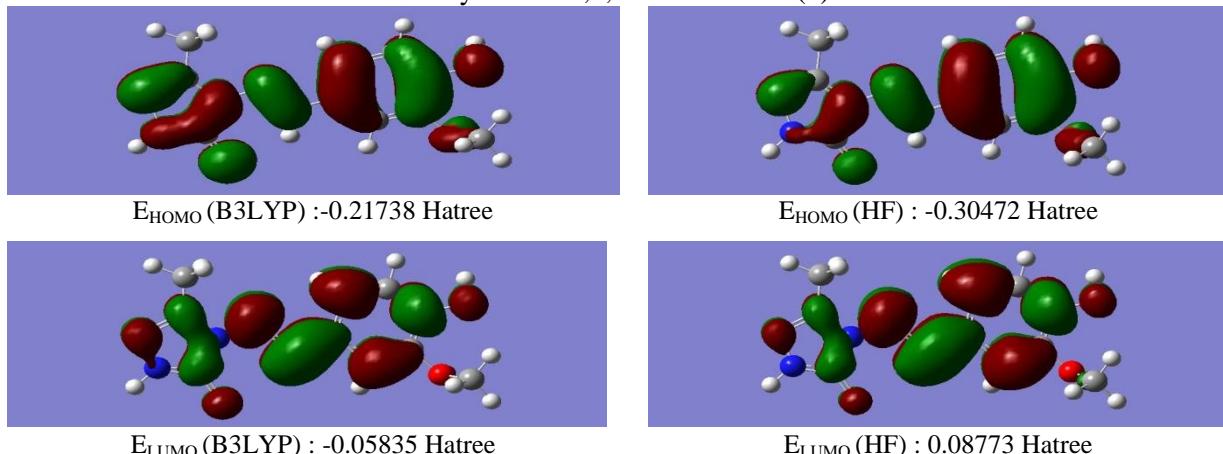
$\lambda$ (nm)B3LYP/HF	Excitation Energy (eV) B3LYP/HF	$f$ (oscillator strengths) B3LYP/HF
313.74/240.30	3.9518/5.1596	0.6044/0.5670
283.46/218.12	4.3740/5.6841	0.0092/0.0141
270.86/191.49	4.5774/6.4749	0.0700/0.0059

**Fig.9.** The calculated absorption wavelength ( $\lambda$ ), excitation energies and oscillator strengths ( $f$ ) and UV-vis spectrums (B3LYP/HF) of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1**).

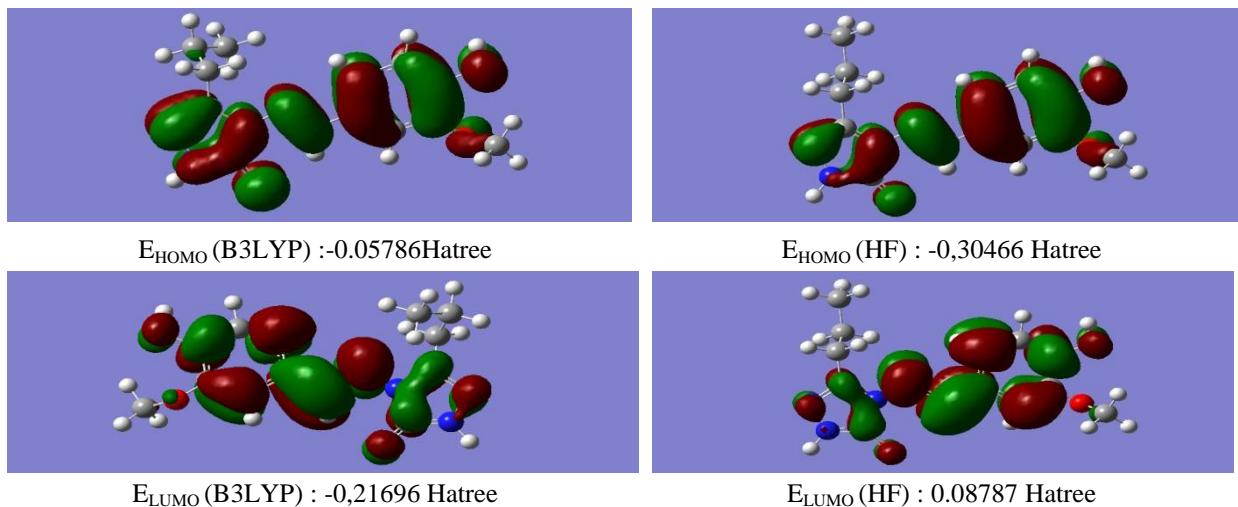


$\lambda$ (nm)B3LYP/HF	Excitation Energy (eV) B3LYP/HF	$f$ (oscillator strengths) B3LYP/HF
314.16/223.27	3.9466/5.5532	0.5643/0.4944
284.08/242.45	4.3644/5.6756	0.0261/0.0256
271.44/194.15	4.5677/6.0143	0.0730/0.0092

**Fig. 10.** The calculated absorption wavelength ( $\lambda$ ), excitation energies and oscillator strengths ( $f$ ) and UV-vis spectrums (B3LYP/HF) of 3-*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one (**2**).



**Fig.11.** 3D plots of HOMO and LUMO energies of 3-methyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1**) at the B3LYP/HF 6-311G(d) level.



**Fig. 12.** 3D plots of HOMO and LUMO energies of 3-n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**2**) at the B3LYP/HF 6-311G(d) level.

### 3.5 Atomic Charges, Thermodynamic Properties and Dipole Moments

The calculated Mulliken atomic charges [9] and results of thermodynamic parameter by using B3LYP/HF 6-311G(d) methods of compounds **1** and **2** in gase phase are listed in (Table 7 and 8), respectively. The electronegative N24, N25, N26, N27, O28, O29, O30 and N30, N31, N32, N33, O34, O35, O36 atoms have negative atomic charge values. The atomic charges of these atoms were calculated as -0.313/-0.380, -0.221/-0.286, -0.364/-0.469, -0.215/-0.277, -0.393/-0.530, -0.366/-0.487, -0.357/-0.447 and -0.312/-0.379, -0.226/-0.276, -0.392/-0.531, -0.366/-0.487, -0.357/-0.447 for compound **1** and **2**, respectively. The C1, C2, C3, C6, C7 atoms surrounded with electronegative atoms have positive atomic charge values for compounds **1** and **2**. These values were found as 0.294/0.396 and 0.342/0.450 (C1) 0.534/0.728 and 0.532/0.723 (C2), 0.138/0.241 and 0.128/0.242 (C3) 0.134/0.217 and 0.137/0.217 (C6), 0.162/0.238 and 0.163/0.239 (C7) a.u. for compounds **1** and **2**, respectively. The C2 atom which is surrounded with three electronegative atoms (N, N, O), C1 atom surrounded with two electronegative atoms (N, N) have the highest positive charges values. All hydrogen atoms of these compounds **1** and **2** have positive atomic charge values. Total energy values and dipole moments of above mentioned compounds were calculated by using B3LYP/HF 6-311G(d,p) methods. The calculated energy values and dipole moment values are given in Table 7 and 8.

**Table 7.** The calculated thermodynamic properties of 3-methyl/n-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1H-1,2,4-triazol-5-one molecule (**1** and **2**)

Parameter	Value Compound 1	Value Compound 2
<b>Thermal energy, E (cal/mol K)</b>	<b>Dft/Hf</b>	<b>Dft/Hf</b>
Elektronic	0.000/0.000	0.000/0.000
Transnational	0.889/0.889	0.889/0.889
Rotational	0.889/0.889	0.889/0.889
Vibrational	152.989/163.499	185.478/203.087
Total	154.767/165.277	187.256/204.864
<b>Dipole Moment D (Debye)</b>	<b>Dft/Hf</b>	<b>Dft/Hf</b>
$\mu_x$	1.7702/1.424	0.565/-0.941
$\mu_y$	3.3397/3.607	4.134/3.579
$\mu_z$	1.3008/1.540	1.720/-1.811
$\mu_{\text{Toplam}}$	3.9974/4.1730	4.518/4.120
<b>Zero-Point Vibrational energy (kcal/mol)</b>	<b>Dft/Hf</b>	<b>Dft/Hf</b>
	144.18425 /155.23268	178.37675/193.19008
<b>Sum of electronic and zero-point Energy</b>	<b>Dft/Hf</b>	<b>Dft/Hf</b>
	-871.024275/-865.828355	-949.495778/-943.855302

<b>(Hartree/Particle)</b>		
<b>Sum of electronic and thermal Energies (Hartree/Particle)</b>	<b>Dft/Hf</b> -871.007411/-865.812349	<b>Dft/Hf</b> -949.481628/-943.836698
<b>Sum of electronic and thermal Enthalpies (Hartree/Particle)</b>	<b>Dft/Hf</b> -871.006467/-865.811405	<b>Dft/Hf</b> -949.480684/-949.480684
<b>Sum of electronic and thermal Free Energies (Hartree/Particle)</b>	<b>Dft/Hf</b> -871.069886/-871.069886	<b>Dft/Hf</b> -949.537383/-949.537383
<b>Rotational constants (GHZ)</b>		
A	<b>Dft/Hf</b> 1.1547317	<b>Dft/Hf</b> 0.6982621
B	<b>Dft/Hf</b> 0.1680958	<b>Dft/Hf</b> 0.1589898
C	<b>Dft/Hf</b> 0.1492125	<b>Dft/Hf</b> 0.1328035
<b>B3LYP:</b> -871.25404745 a.u.		
<b>HF:</b> -866.07573362 a.u.		

**Table 8.**The Mulliken atomic charges of 3-methyl/*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule (**1** and **2**)

Atoms <b>Compound 1</b>	Charges (a.u.)		Atoms <b>Compound 2</b>	Charges (a.u.)	
	<b>DFT</b>	<b>HF</b>		<b>DFT</b>	<b>HF</b>
<b>C1</b>	0.294	0.396	<b>C1</b>	0,342	0,450
<b>C2</b>	0.534	0.728	<b>C2</b>	0,532	0,723
<b>C3</b>	0.138	0.241	<b>C3</b>	0,128	0,242
<b>C4</b>	-0.158	-0.181	<b>C4</b>	-0,163	-0,186
<b>C5</b>	-0.048	-0.076	<b>C5</b>	-0,048	-0,076
<b>C6</b>	0.134	0.217	<b>C6</b>	0,134	0,217
<b>C7</b>	0.162	0.238	<b>C7</b>	0,163	0,239
<b>C8</b>	-0.104	-0.10	<b>C8</b>	-0,107	-0,106
<b>C9</b>	-0.024	-0.039	<b>C9</b>	-0,019	-0,037
<b>C10</b>	-0.245	-0.182	<b>C10</b>	-0,201	-0,153
<b>C11</b>	-0.115	-0.003	<b>C11</b>	-0,231	-0,204
<b>H12</b>	0.249	0.250	<b>C12</b>	-0,291	-0,242
<b>H13</b>	0.141	0.165	<b>C13</b>	-0,114	-0,003
<b>H14</b>	0.102	0.107	<b>H14</b>	0,249	0,259
<b>H15</b>	0.092	0.098	<b>H15</b>	0,142	0,166
<b>H16</b>	0.100	0.111	<b>H16</b>	0,102	0,107
<b>H17</b>	0.132	0.122	<b>H17</b>	0,091	0,097
<b>H18</b>	0.132	0.122	<b>H18</b>	0,107	0,114
<b>H19</b>	0.132	0.127	<b>H19</b>	0,135	0,128
<b>H20</b>	0.117	0.097	<b>H20</b>	0,141	0,126
<b>H21</b>	0.119	0.099	<b>H21</b>	0,121	0,116
<b>H22</b>	0.094	0.072	<b>H22</b>	0,109	0,104
<b>H23</b>	0.253	0.262	<b>H23</b>	0,110	0,116
<b>N24</b>	-0.313	-0.380	<b>H24</b>	0,108	0,104
<b>N25</b>	-0.221	-0.286	<b>H25</b>	0,110	0,090
<b>N26</b>	-0.364	-0.469	<b>H26</b>	0,094	0,087
<b>N27</b>	-0.215	-0.277	<b>H27</b>	0,117	0,098
<b>O28</b>	-0.393	-0.530	<b>H28</b>	0,119	0,073
<b>O29</b>	-0.366	-0.487	<b>H29</b>	0,253	0,099

Atoms Compound 1	Charges (a.u.)		Atoms Compound 2	Charges (a.u.)	
	DFT	HF		DFT	HF
<b>O30</b>	-0.357	-0.447	<b>N30</b>	-0.312	-0.379
			<b>N31</b>	-0.226	-0.284
			<b>N32</b>	-0.373	-0.482
			<b>N33</b>	-0.211	-0.276
			<b>O34</b>	-0.392	-0.531
			<b>O35</b>	-0.366	-0.487
			<b>O36</b>	-0.357	-0.447

#### 4 CONCLUSION

The molecular structure, vibrational wavenumbers, the electronic absorption maximum wavelenghts, the HOMO, LUMO, atomic charges and thermodynamic parameters of the synthesized 3-methyl/*n*-propyl-4-(3-methoxy-4-hydroxybenzylidenamino)-4,5-dihydro-1*H*-1,2,4-triazol-5-one molecule have been calculated by using HF/6-311G(d,p) and DFT/6-311G(d,p) levels. The calculated theoretical vibrational frequencies and UV spectroscopic parameters are consistent with the experimental data.

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